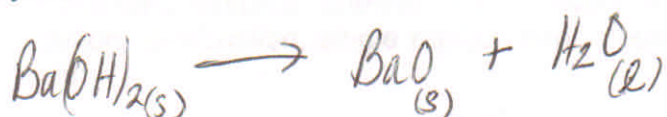


COMPLETING REACTION EQUATIONS

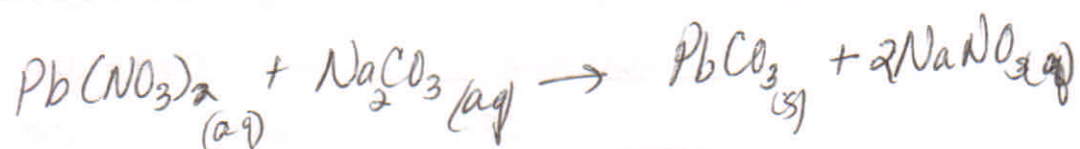
Name _____

Write complete balanced molecular equations for each of the following reactions. use appropriate symbols to show the state of the reactants and products whenever possible.

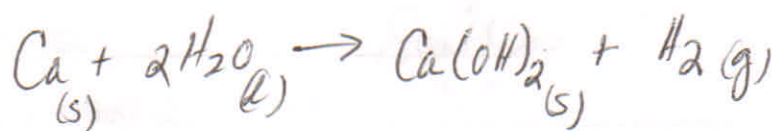
- 1) Solid barium hydroxide is decomposed by heating.



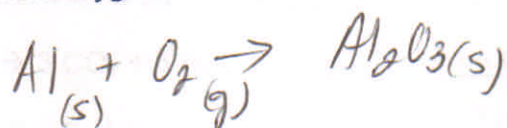
- 2) Aqueous lead (II) nitrate reacts with sodium carbonate solution.



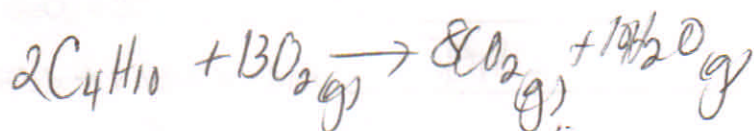
- 3) Solid calcium reacts with water.



- 4) Aluminum reacts with oxygen.



- 5) Butane (C_4H_{10}) burns



- 6) Potassium chlorate decomposes into potassium chloride and oxygen gas.

